

*Crysta'Days | November 2023*

# Single crystal formation in core-shell capsules



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Prof. Dr. Johan Wouters  
Dr. Jean Septavaux

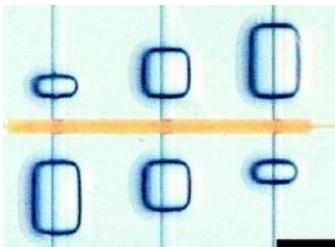


# Single crystal generation

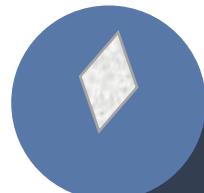
Microfluidic based methods : control at the nanoliter scale



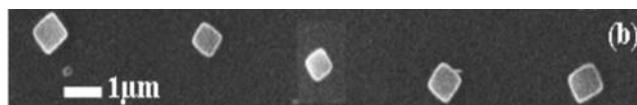
Free interface diffusion in crystallization **chambers**



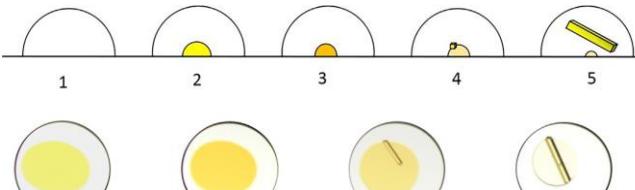
Hansen et al., PNAS, 2002, 99, 26, 16531-16536



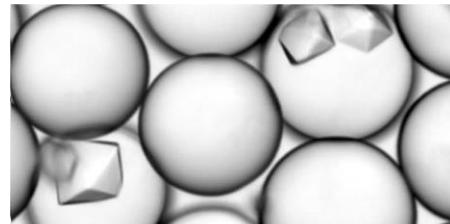
Diffusion in liquid droplets



Grossier et al., APPLIED PHYSICS LETTERS 98, 1 2011



Tyler et al., Chem 6, 2020, 1755-1765



Babnigg et al., Acta Cryst. 2022, D89, 997-1009

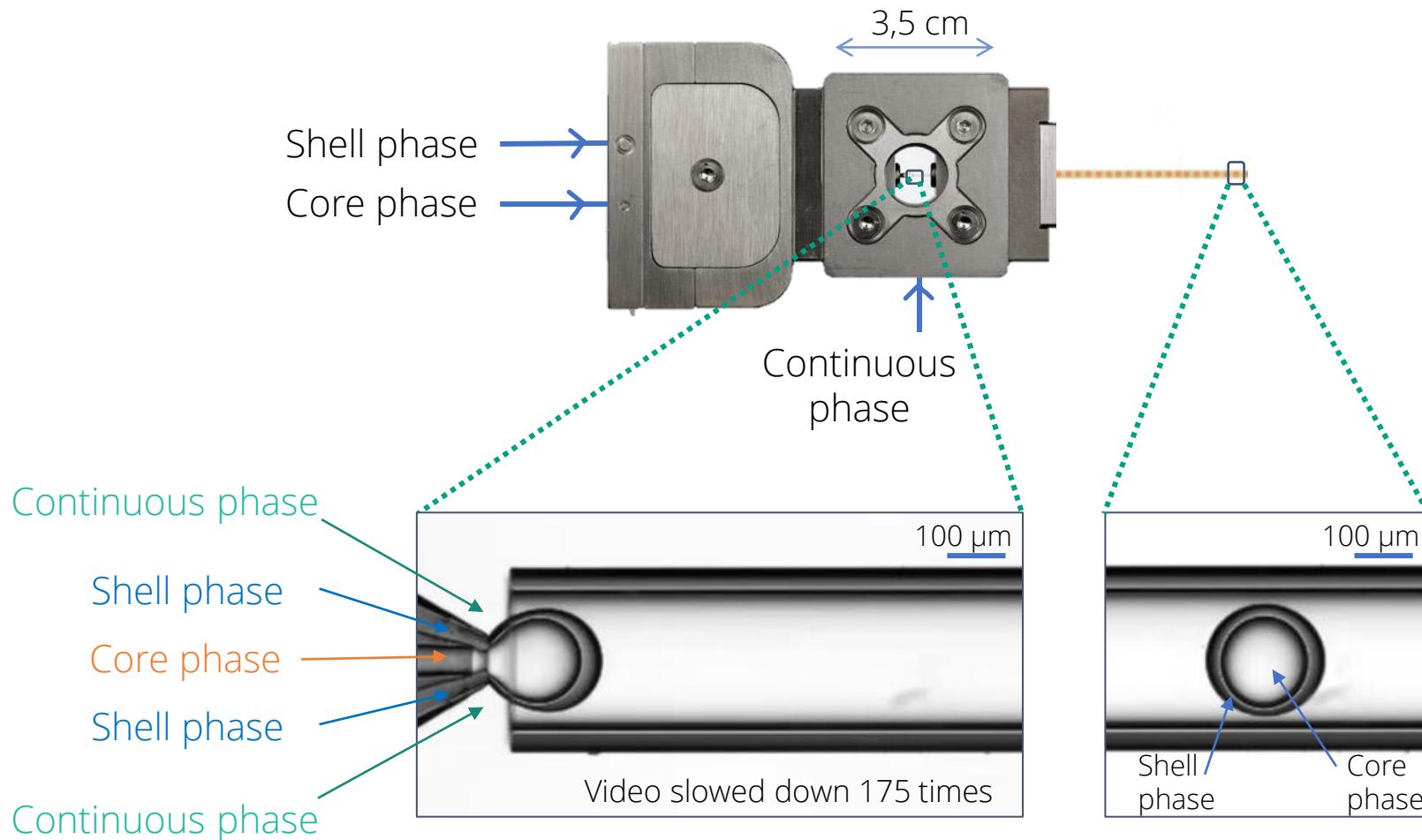


Solid **microcapsules** with aqueous core

- Stability of the emulsion
- Large choice of solvents

# Presentation of the system

An emulsification generator for double emulsions production

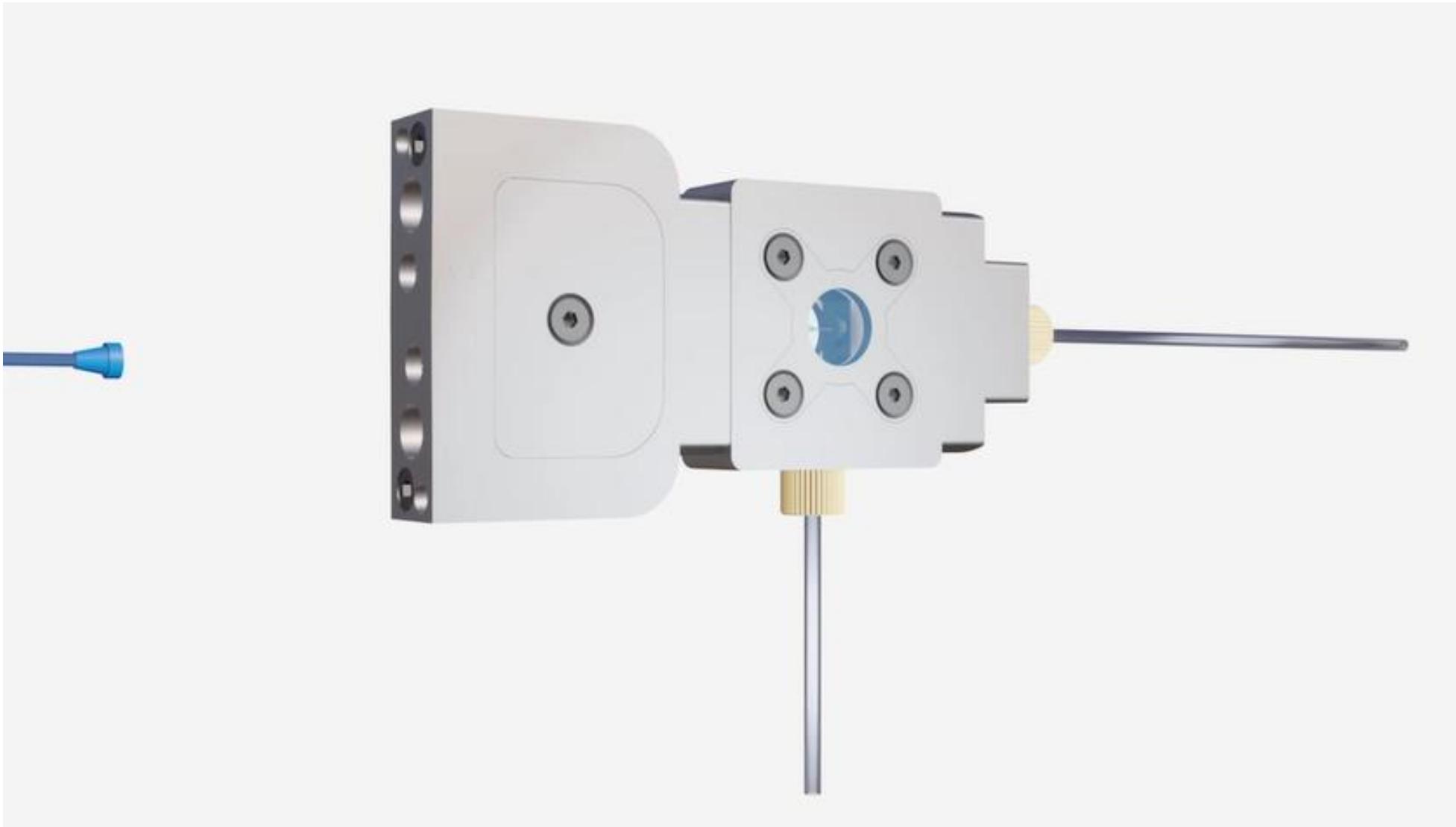


- Single emulsions
- Double emulsions
- No surface treatment needed
- Water/oil/water
- Oil/water/oil
- Highly monodispersed droplets
- Frequency > 100 Hz
- Size range: 40 to 400 µm

In this work, the diameter size of capsules is between 237 and 309 µm.

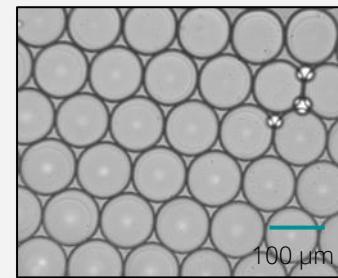
# Presentation of the system

The Raydrop microfluidic droplet generator

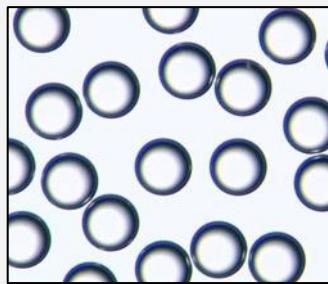


<https://youtu.be/vpbdwMb5mM0?si=sOOYyj8ap7uBxfW1>

# Variety of emulsions

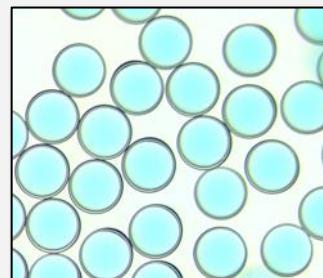


Alginate

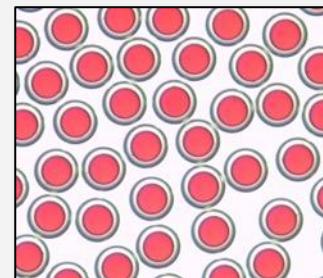


Eudragit™

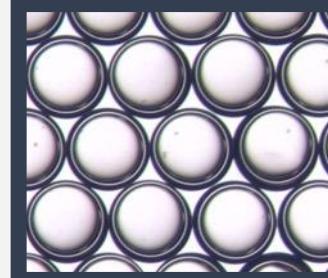
## Double emulsions and microcapsules



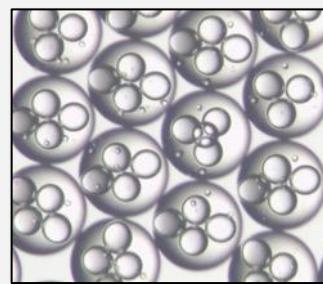
PLGA



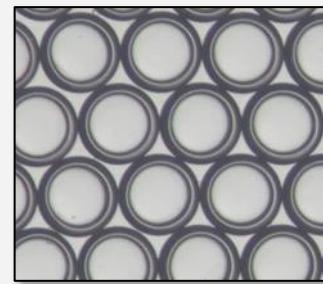
Chitosan



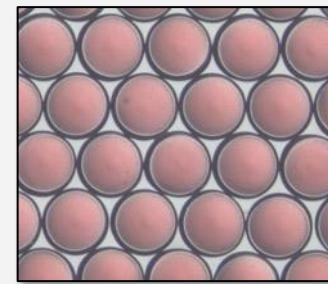
Polymethacrylate  
(aqueous core)



Multiple emulsion



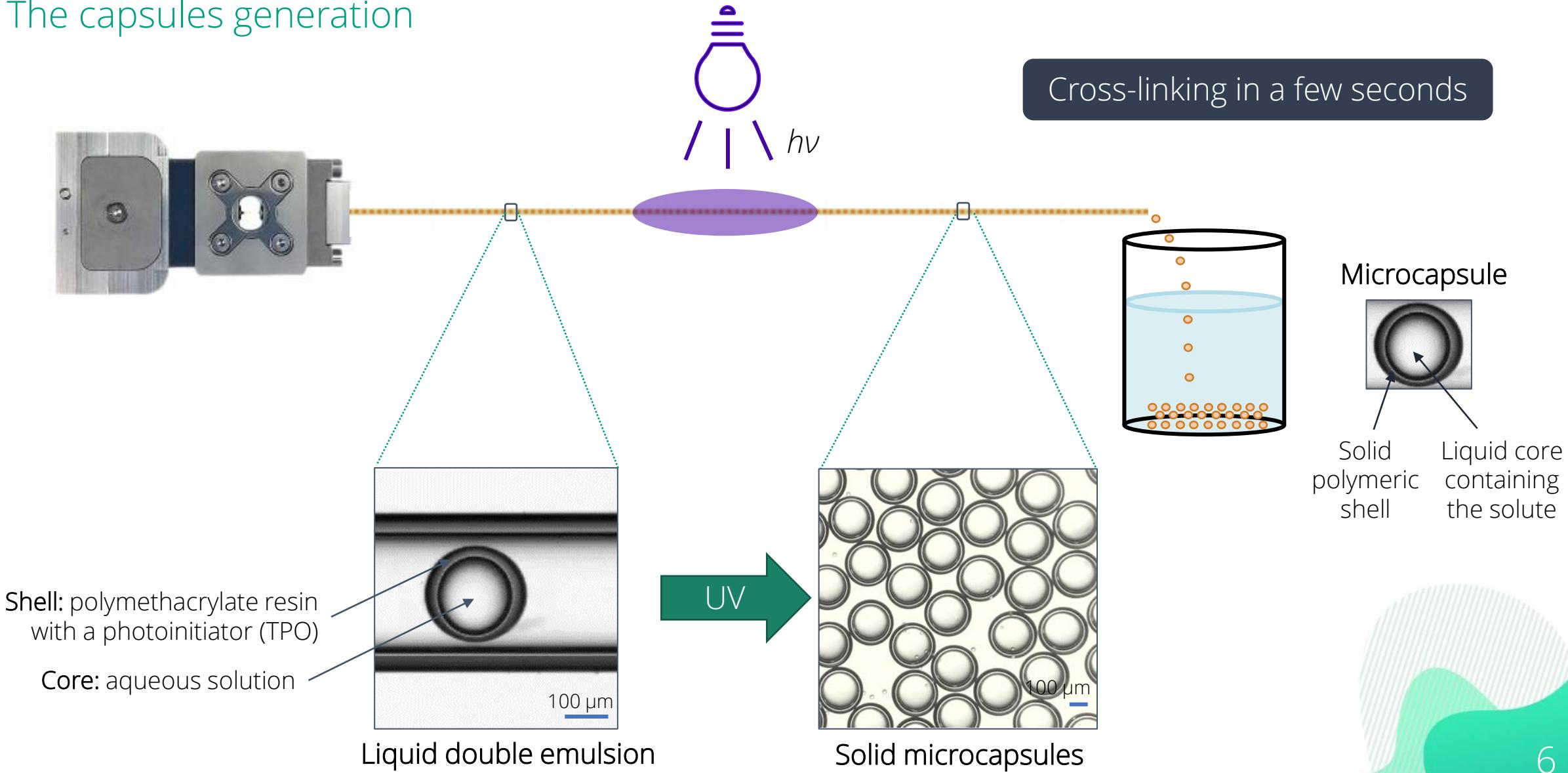
PEGDA



Polymethacrylate  
(oily core)

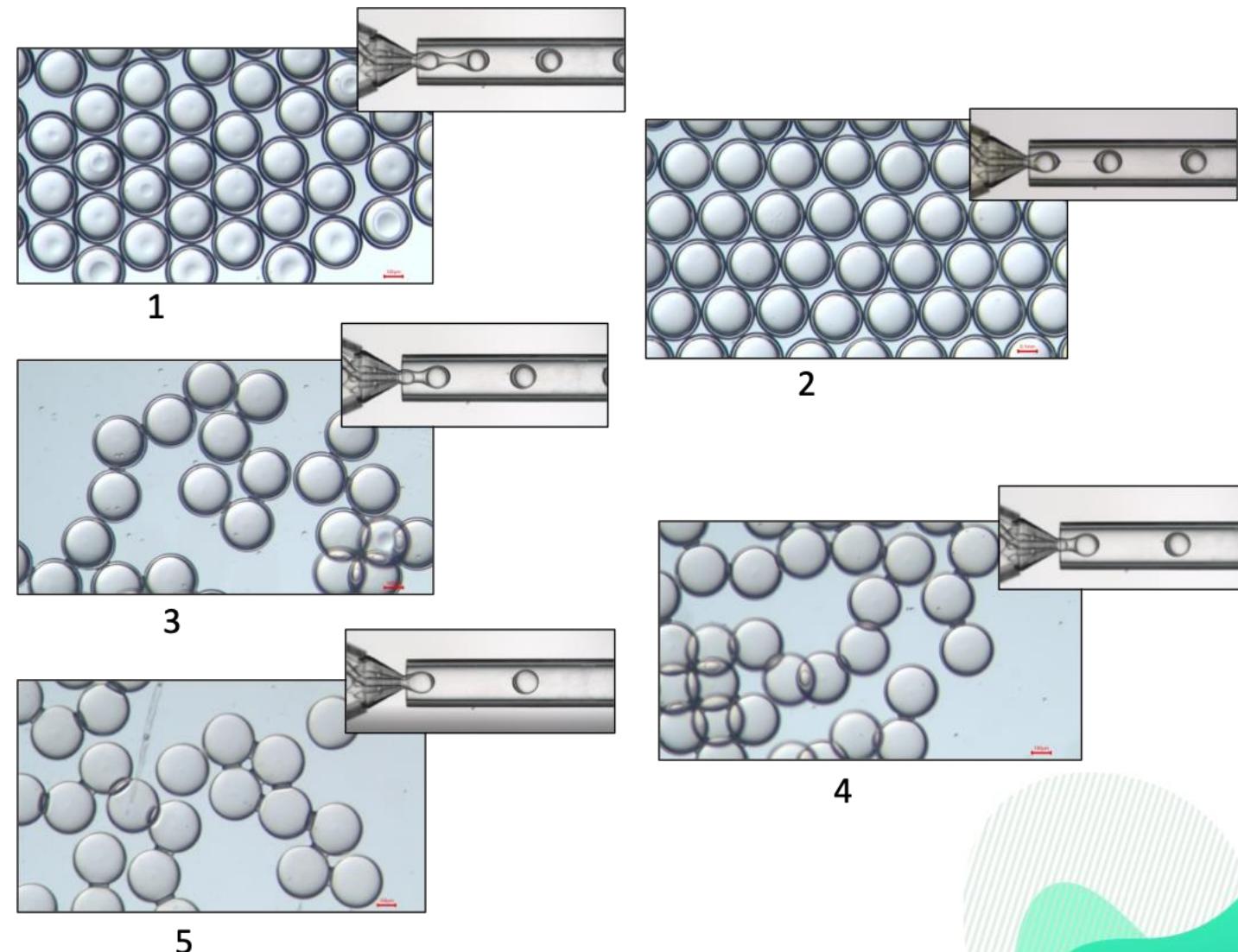
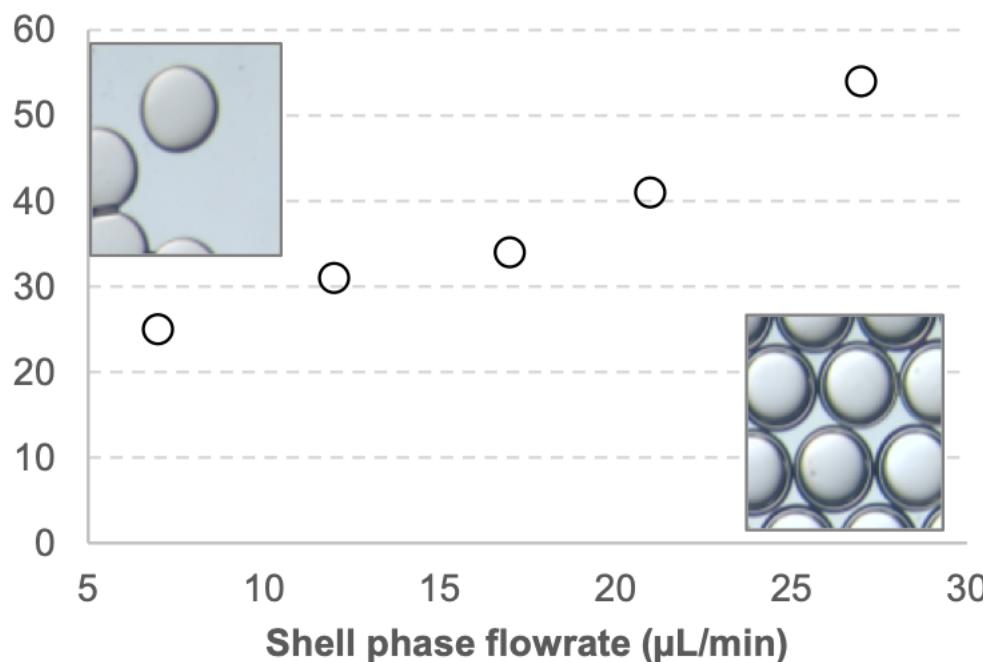
# How to create UV-cured capsules?

The capsules generation



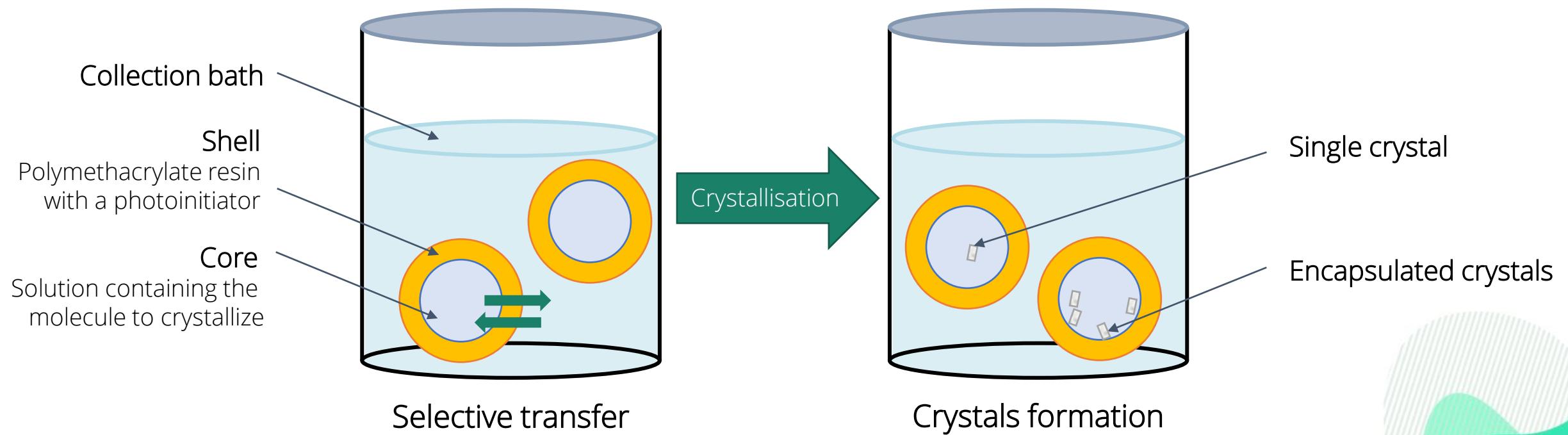
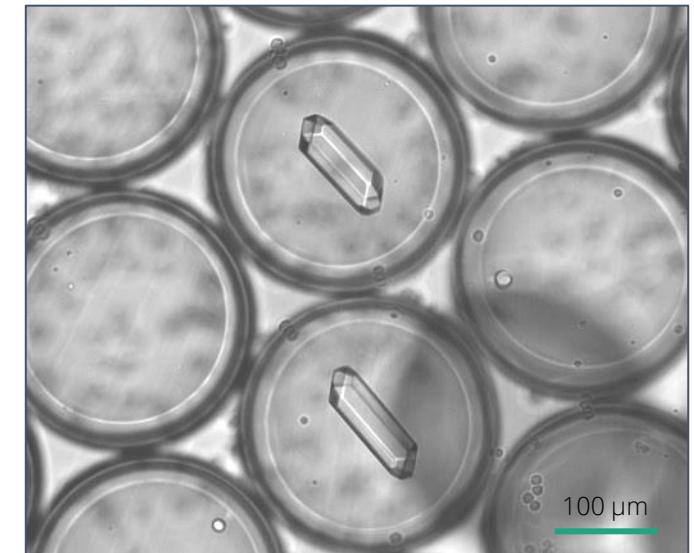
# Adjustment of the shell thickness

Various shell thickness



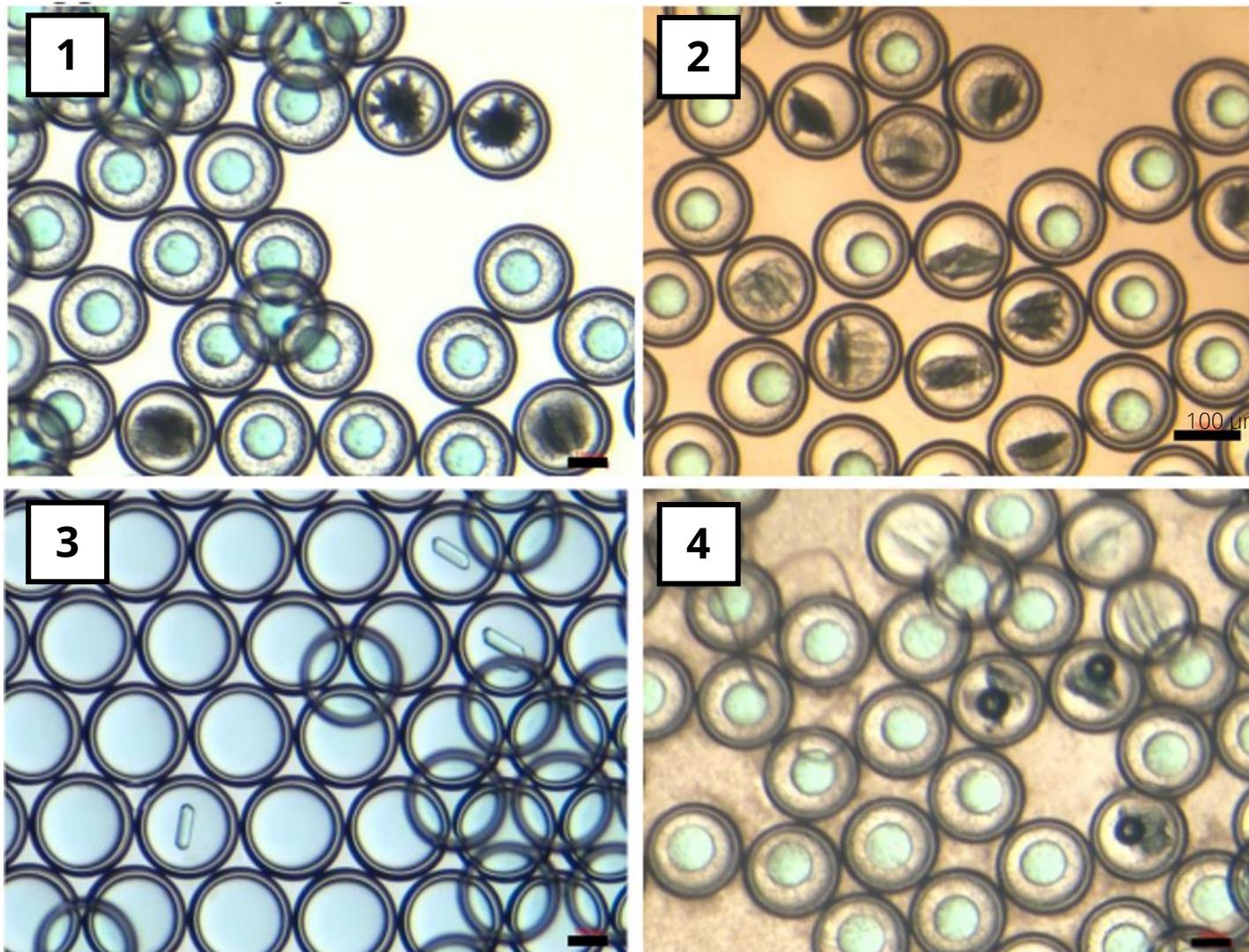
# Presentation of the system

## Crystal formation



# Results – crystallized molecules

Small molecules crystals - Copper (II) sulfate at 200 g/L in water

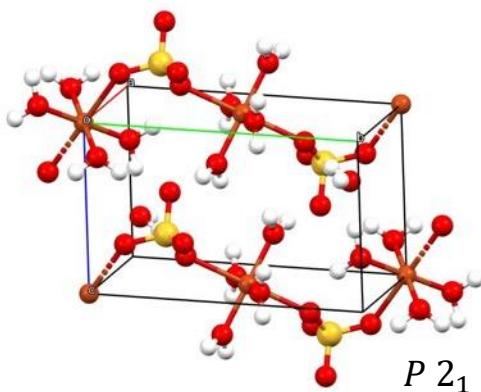
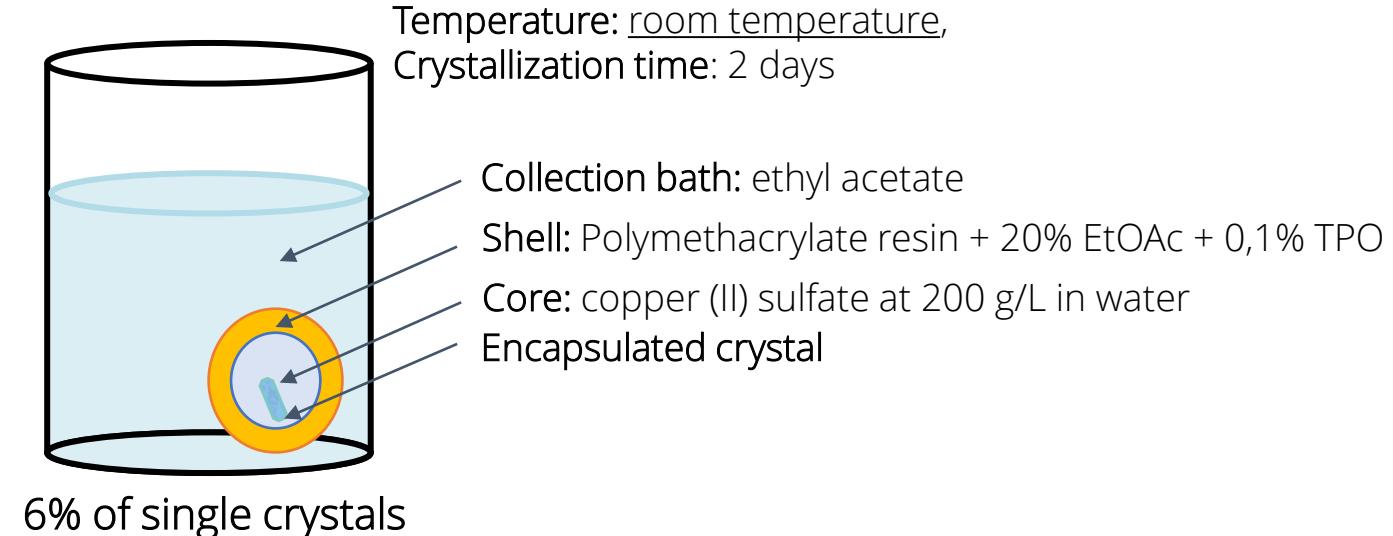
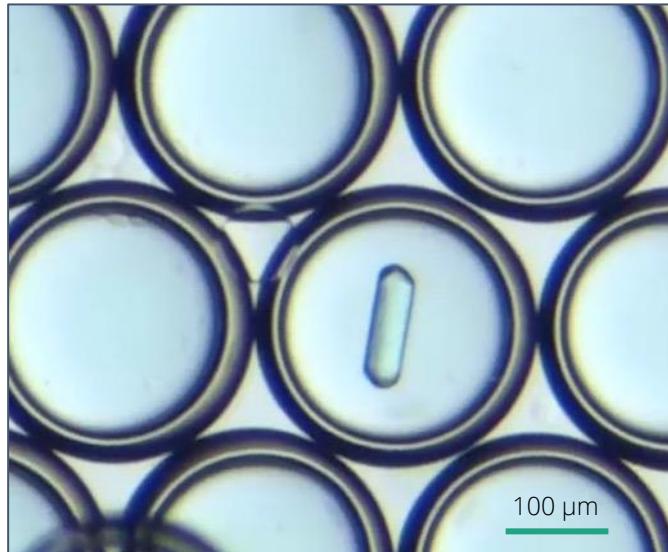


Screening parameters:

| Collection bath      | Storage      |        |        | Picture |
|----------------------|--------------|--------|--------|---------|
| Air                  | 22 °C        | 4.5 °C | -18 °C |         |
| Water                | 22 °C        | 4.5 °C | -18 °C |         |
| <b>Acetone</b>       | <b>22 °C</b> | 4.5 °C | -18 °C | 2       |
| Methanol             | 22 °C        | 4.5 °C | -18 °C |         |
| <b>Ethanol</b>       | <b>22 °C</b> | 4.5 °C | -18 °C | 1       |
| <b>2-propanol</b>    | <b>22 °C</b> | 4.5 °C | -18 °C | 4       |
| <b>Ethyl acetate</b> | <b>22 °C</b> | 4.5 °C | -18 °C | 3       |

# Results – crystallized molecules

Small molecules crystals - Copper (II) sulfate



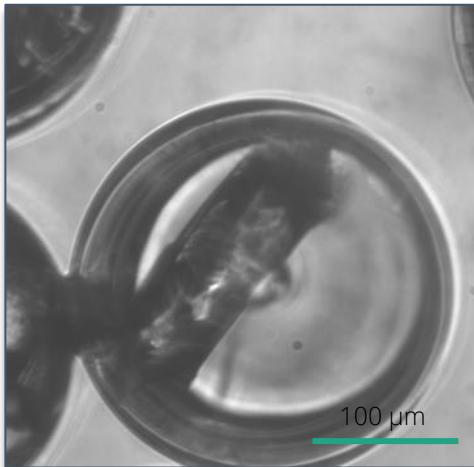
Crystal structure of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

Crystallization by antisolvent diffusion

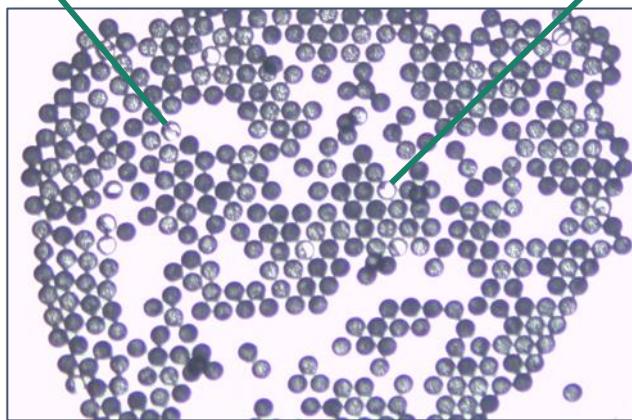
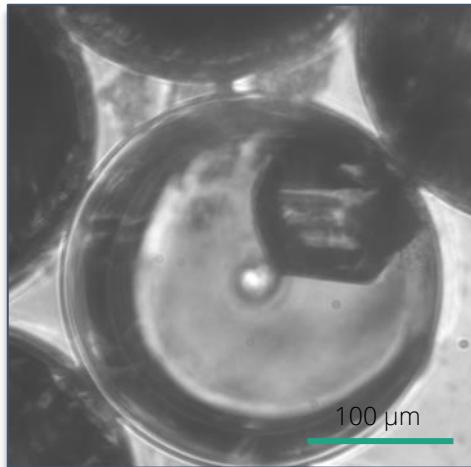
# Results – crystallized molecules

## Polymorphism - Glycine

Alpha

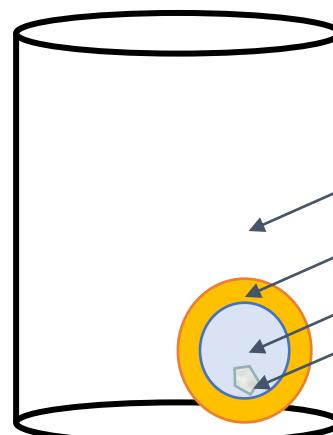


Gamma



Sample of 500 capsules

← 10 seconds of production



Temperature: 4.5°C  
Crystallization time: 1 day

Collection bath: air

Shell: Polymethacrylate resin + 20% EtOAc + 0,1% TPO

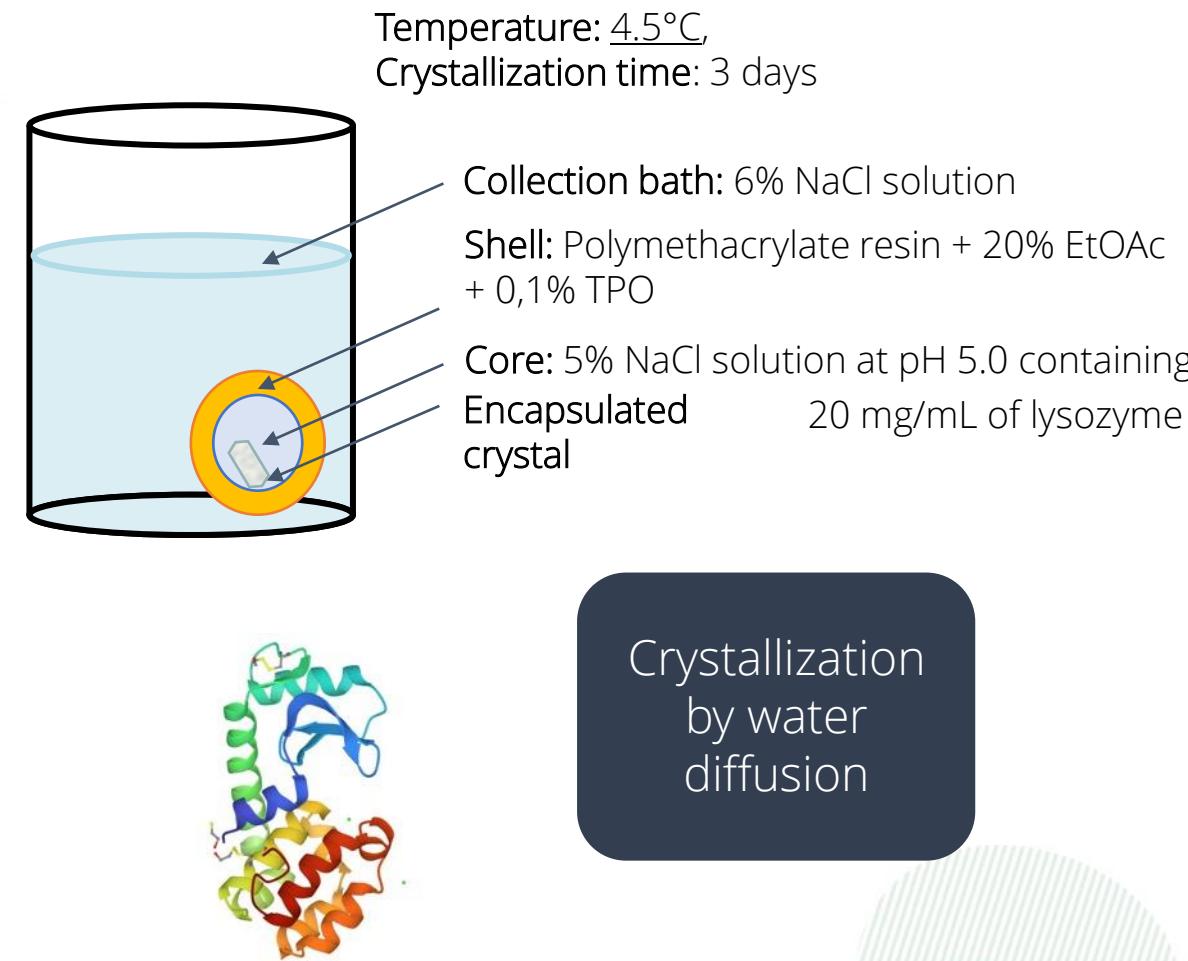
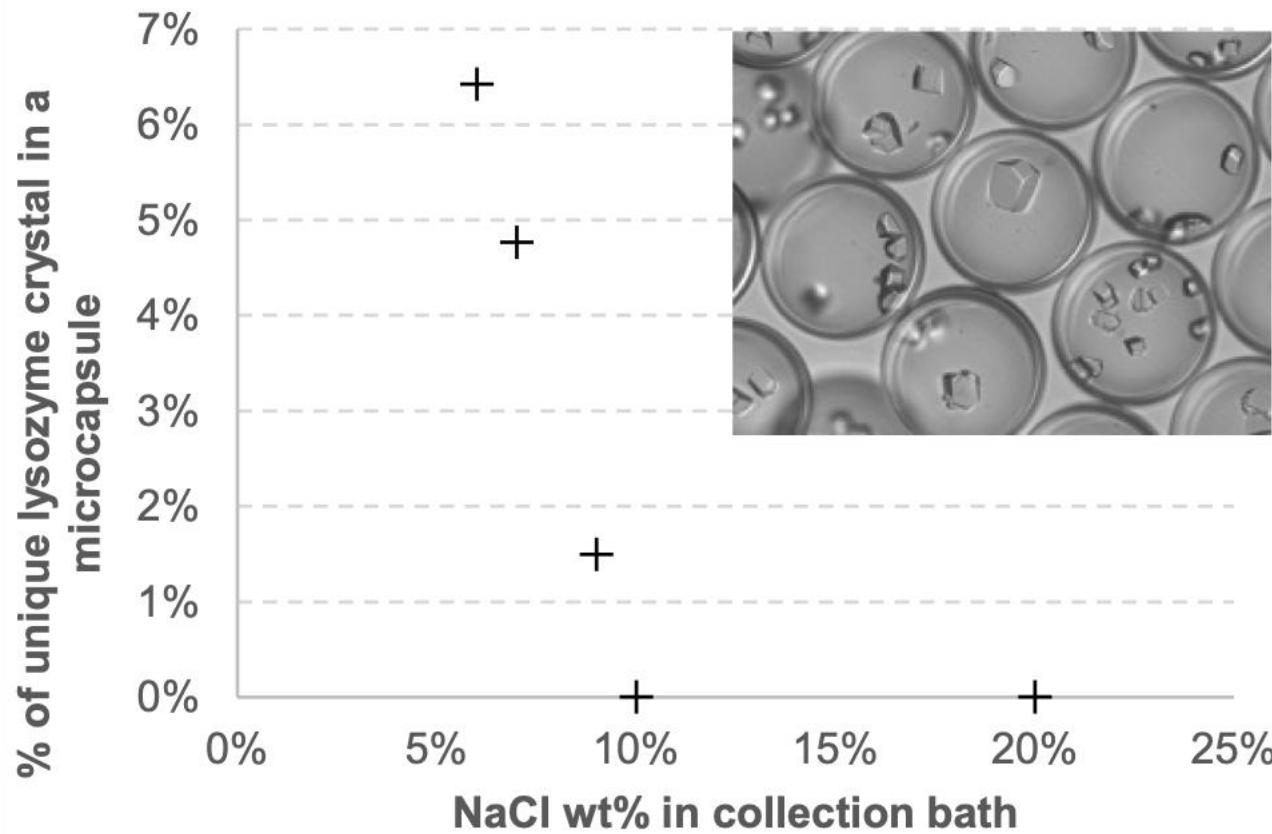
Core: glycine at 190 g/L in water  
Encapsulated crystal

2.5% of single crystals

Screening of polymorphes of glycine by producing hundreds of capsules

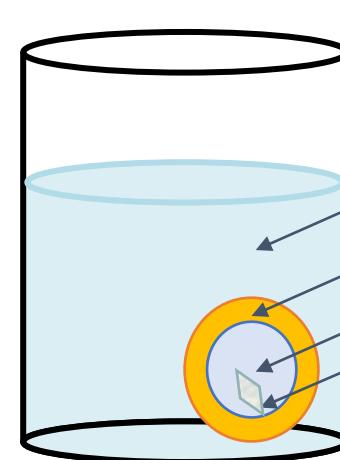
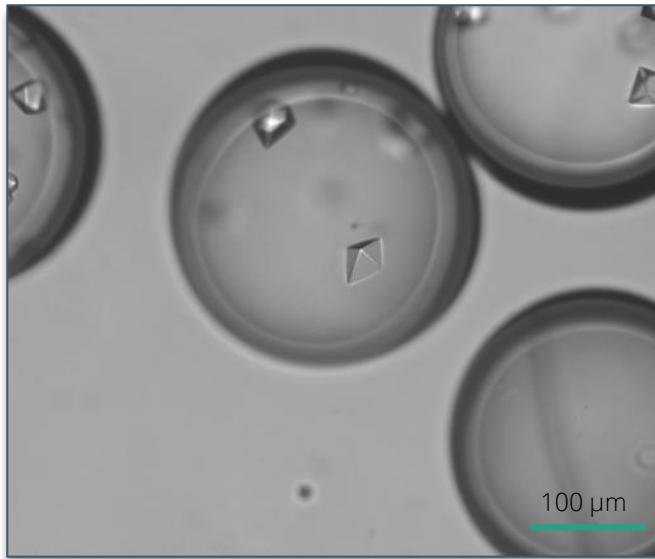
# Results – crystallized molecules

## Protein crystals - Lysozyme



# Results – crystallized molecules

Protein crystals - Proteinase K

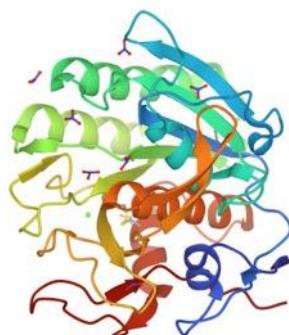


Temperature: 4.5°C,  
Crystallization time: 2 days

Collection bath: 1.2M ammonium sulfate solution

Shell: Polymethacrylate resin + 20% EtOAc + 0,1% TPO

Core: 1.1M ammonium sulfate solution at pH 7.0 containing  
20 mg/mL of proteinase K

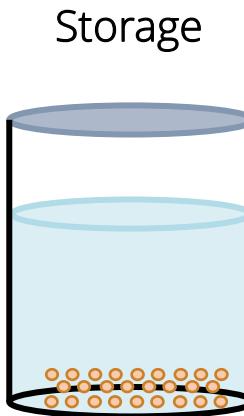


Crystal structure of proteinase K  
29 kDa

Crystallization by water diffusion

# Direct crystallography

Collect of XRD data

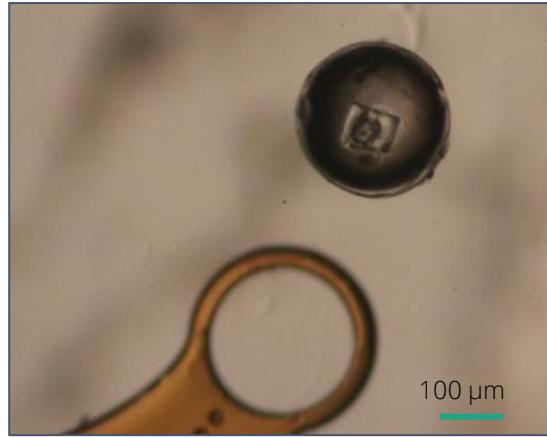


Storage

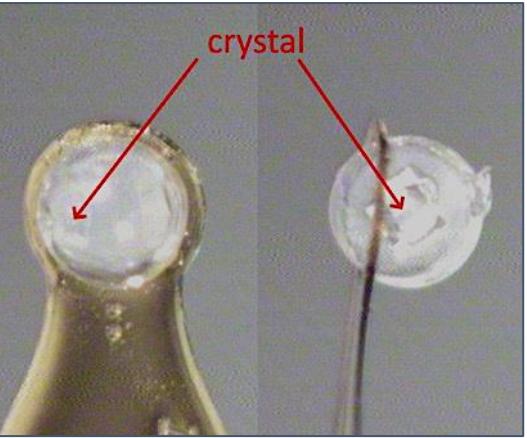
Shipment



XRD analysis



Mounting loop: 300  $\mu\text{m}$  aperture



In situ

Ex situ

Capsules can accept :

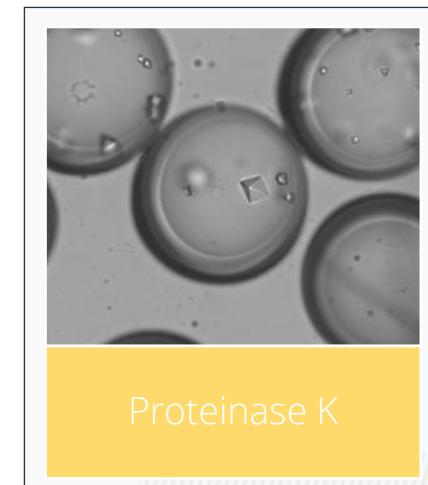
- Room temperature
- Low temperature (up to -196°C with liquid nitrogen)

without degradation of the shell.



# Summary

- High throughput  
≈2500 independent capsules/minute
- Not resource consuming  
Down to 50 ng solute/capsule,  
50 µg of solute needed for one experiment
- Easy screening  
Fast parameter tuning
- Crystals encapsulated in a solid shell  
Crystals protection, compatible with N<sub>2</sub> freezing
- Transparent shell  
Allows direct analysis



# Publication of results

ChemComm



COMMUNICATION

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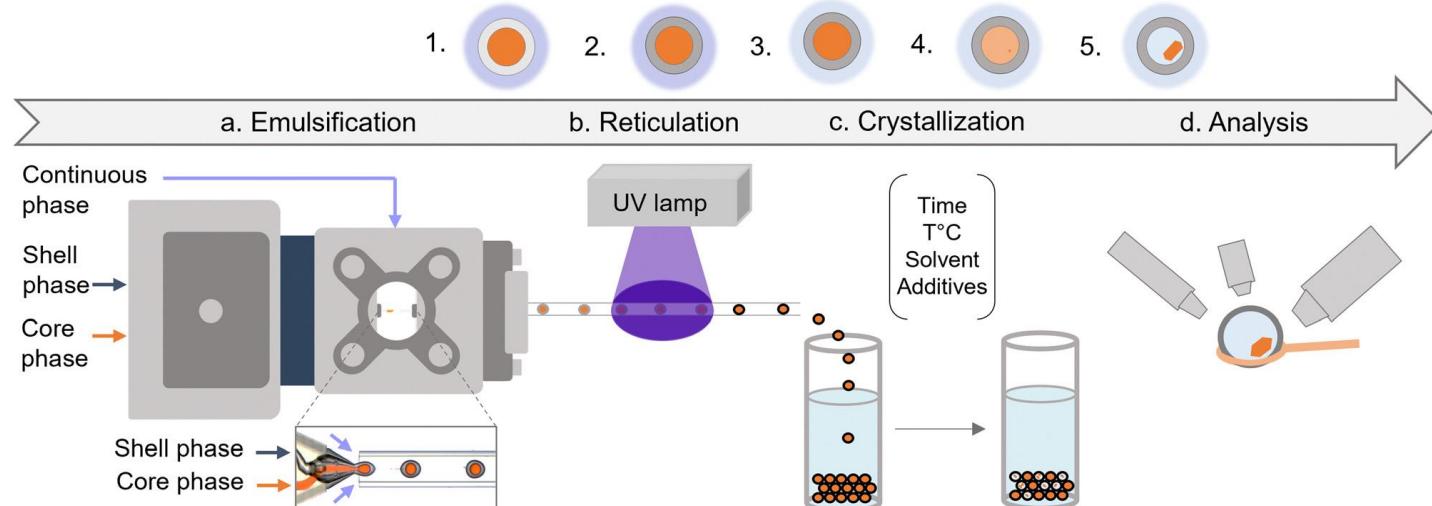
DOI: 10.1039/d3cc03727d

rsc.li/chemcomm

## Single crystal formation in core–shell capsules†

Marie Mettler,<sup>a</sup> Adrien Dewandre,<sup>a</sup> Nikolay Tumanov,<sup>b</sup> Johan Wouters<sup>b</sup> and  
Jean Septavaux<sup>b,\*a</sup>

DOI: 10.1039/d3cc03727c



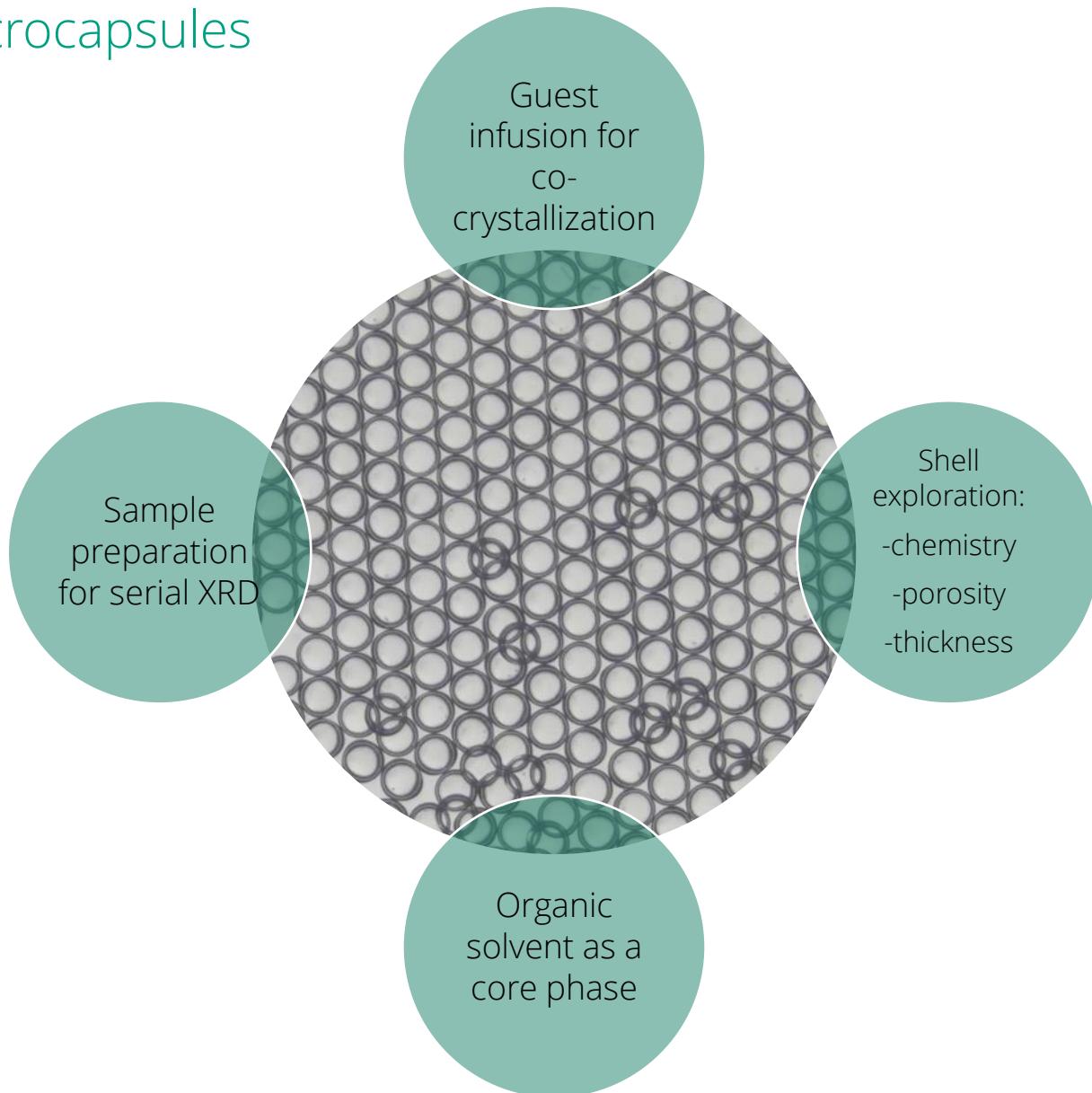
Representation of the crystallization process

Mettler *et al.*, *Chem. Commun.*, 2023, 59, 12739-12742

<https://pubs.rsc.org/en/content/articlelanding/2023/cc/d3cc03727d>

# Perspectives

## Single crystals in microcapsules



# Acknowledgments

The Secoya team

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Bertrand Loriers

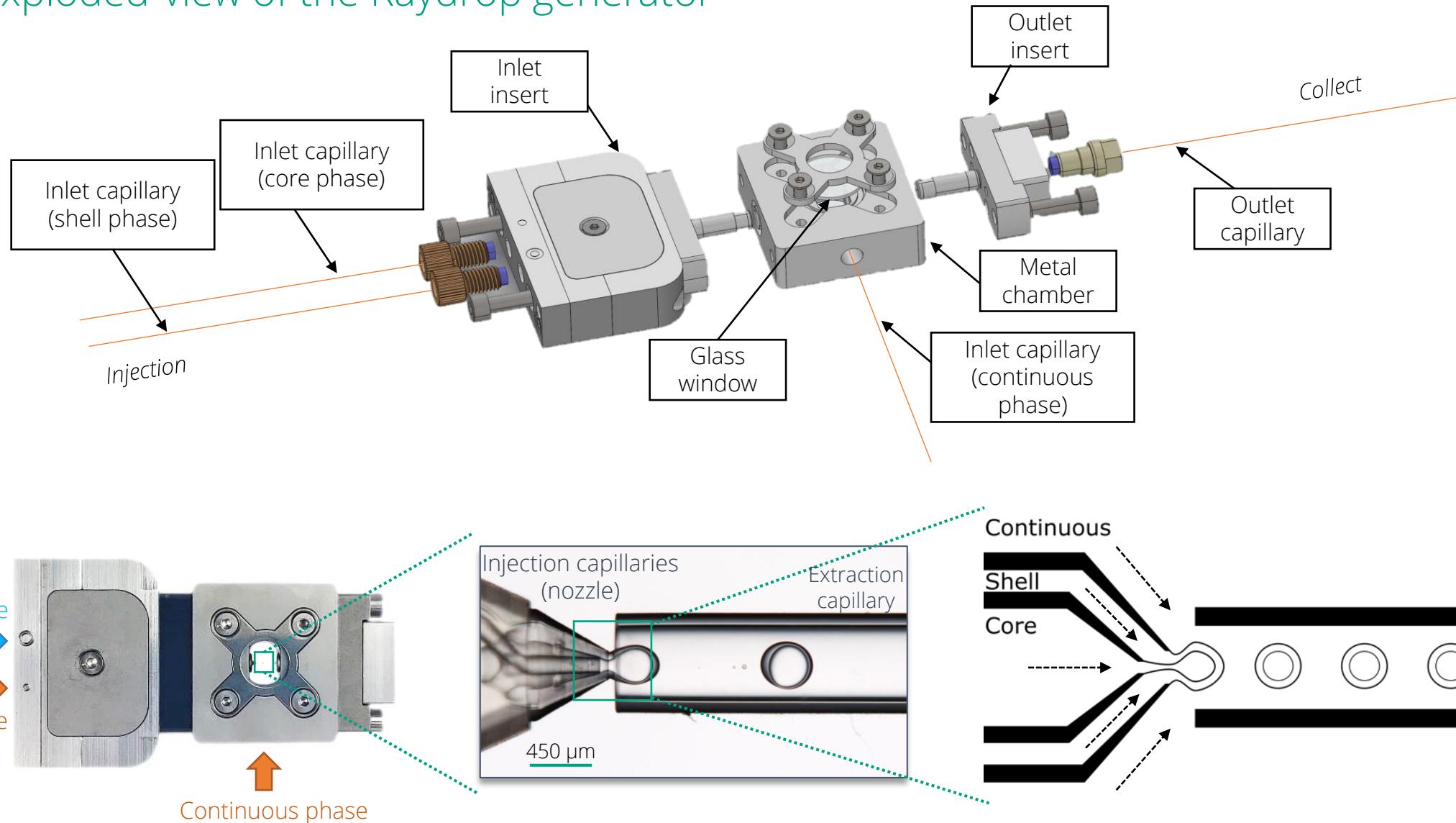
University of Namur

Dr. Nikolay Tumanov  
Prof. Dr. Johan Wouter



# Annexe - Presentation of the system

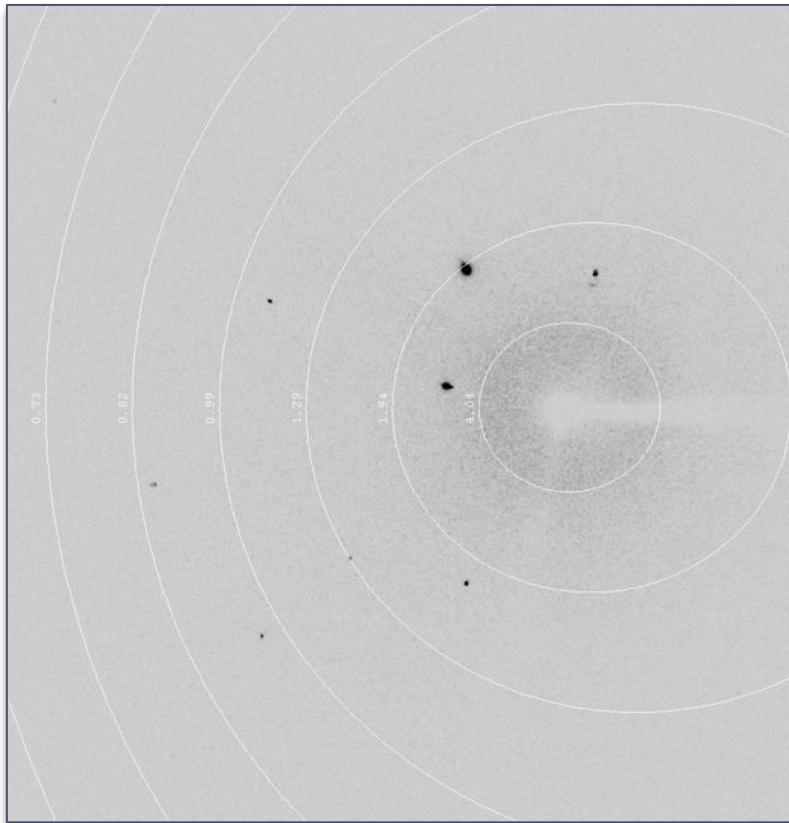
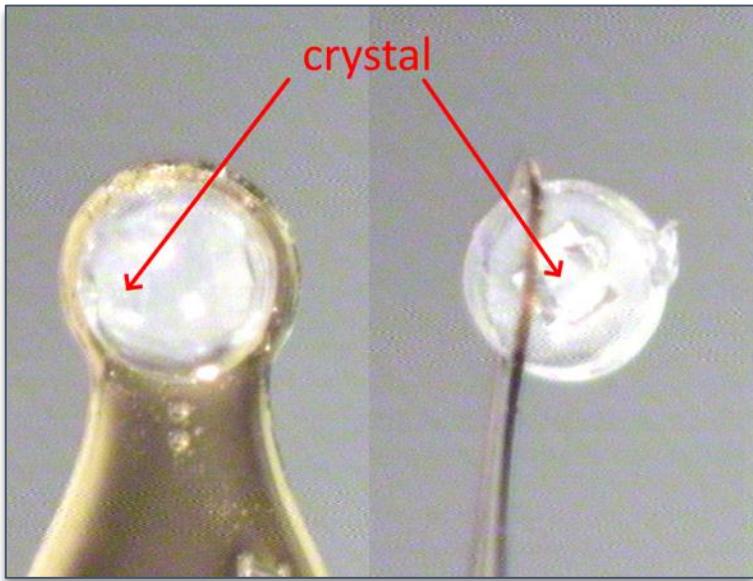
## Exploded-view of the Raydrop generator



# Annexe - Diffractometer

X-ray diffraction study

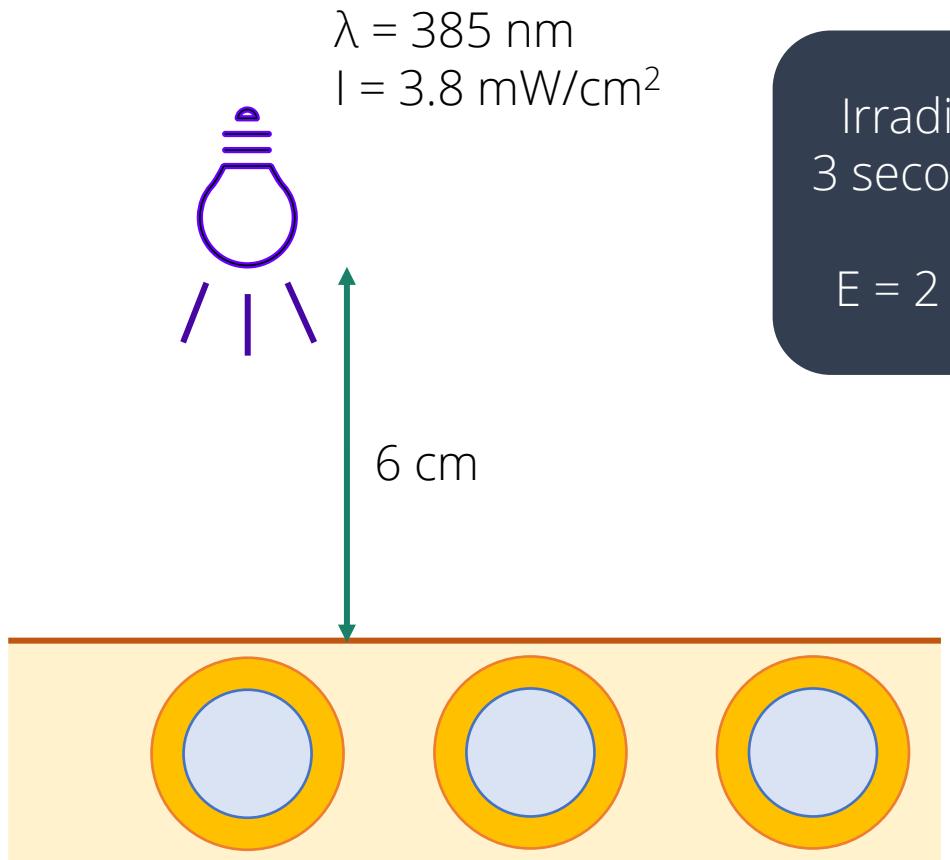
Diffractometer



| Parameter                    | Details for NaCl    |
|------------------------------|---------------------|
| Crystal system               | cubic               |
| Space group                  | Fm-3m               |
| $a/\text{\AA}$               | 5.6357(3)           |
| $b/\text{\AA}$               | 5.6357(3)           |
| $c/\text{\AA}$               | 5.6357(3)           |
| $\alpha/^\circ$              | 90                  |
| $\beta/^\circ$               | 90                  |
| $\gamma/^\circ$              | 90                  |
| Volume/ $\text{\AA}^3$       | 179.00(3)           |
| Crystal size/mm <sup>3</sup> | 0.059 × 0.13 × 0.15 |

# Annexe – Solidification of the shell

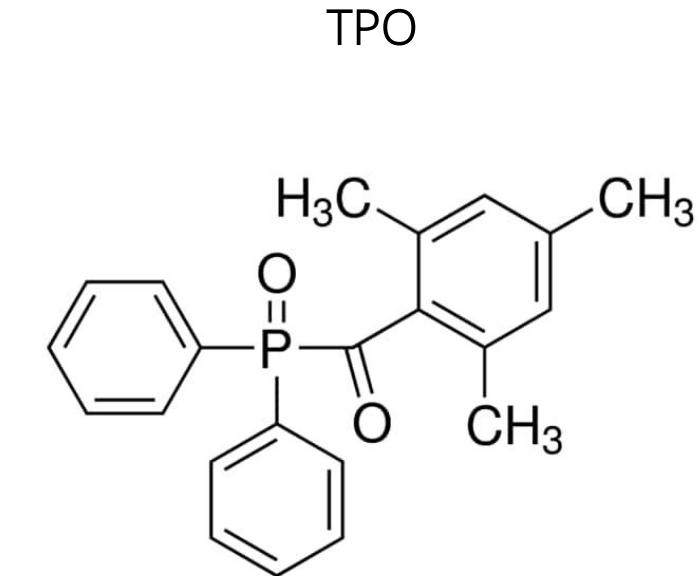
Resin cross-linking process with a photoinitiator



UV exposition in  
the outlet tubing

$$\lambda = 385 \text{ nm}$$
$$I = 3.8 \text{ mW/cm}^2$$

Irradiation time:  
3 seconds/capsule  
 $E = 2 \text{ mJ/capsule}$



Diphenyl(2,4,6-trimethylbenzoyl)phosphine  
oxide

Free-radical photoinitiator